Stimulated by these results, a new gradient furnace has been built. Its main characteristic is that the heat flow is transverse to fibres in dovetail-shaped samples. The temperature distribution is symmetrical and the temperature is carefully controlled. Details are to be published elsewhere [3].

A Co-17% Cr-10% NiTaC eutectic was directionally solidified at a rate of 0.6 cm h^{-1} . 5 mm diameter specimens were annealed in a temperature gradient of 270 K mm⁻¹ in a vacuum in excess of 10^{-1} Pa. After 250 h annealing no change in microstructure was detected, and no increase in the fibre diameter or spacing or any change in fibre shape were observed (Fig. 1).

It is concluded, therefore, that under the above-mentioned test conditions there is no visible

Flux growth of lanthanum borate, LaBO₃

The rare earth borates, RBO_3 , fall into three groups according to their crystal structure, each group being structurally related to one of the three forms of calcite. LaBO₃ has the orthorhombic structure of aragonite and is pseudocoarsening or fibre degradation in this eutectic system.

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hexagonal, with the pseudohexagonal c-axis parallel to the orthorhombic b-axis [1-3].

It has been shown that $LaBO_3$ can be grown from PbO-B₂O₃ in the form of small rods or platy crystals [4]. Recently, a model has been proposed for the prediction of starting compositions for the flux growth of crystals with one

Starting composition* Crucible Cooling Notes on the results Max. temvolume (mol%) (dry perature rate and materials) (ml) and soak minimum period temp. °C h⁻¹ to °C °C La_2O_3 B_2O_3 PbO h 4.040 56.0 10 1250 700 15 3 Many small transparent platelets grew in a layer at the surface 3.7 12 84.3 1250 10 15 3 700 Large transparent platy crystals, up to $10 \text{ mm} \times 3 \text{ mm} \times 1 \text{ mm}$, at the melt surface only 4.3 13.2 82.5 10 1250 15 3 700 All crystals grew at the base of the crucible. Rods up to $3 \text{ mm} \times 1.5 \text{ mm}$ × 1 mm, tabular crystals up to 4 mm \times 3 mm \times 1 mm and equi-dimensional crystals 2 mm on edge 4.6 13.2 82.2 10 1250 15 3 700 Rods up to $4 \text{ mm} \times 1.5 \text{ mm} \times 1 \text{ mm}$ and faceted crystals 3 mm × 2 mm × 2 mm grew at the crucible base. Solution was complete 5.0 13.1 81.9 10 1250 15 3 700 Many intergrown crystals, indicating that solution was not complete

TABLE I Compositions and growth conditons for LaBO₃

*Several batches of each composition were prepared.





Figure 1 (a) Platelets of LaBO₃ (1 mm grid) (b) Tabular prisms of LaBO₃ (1 mm grid) (c) Approximately equidimensional crystals of LaBO₃ (1 mm grid).

refractory and one relatively low-melting component (such as La_2O_3 and B_2O_3 , respectively) from fluxed melts [5]. According to this model, increasing the proportion of basic oxide relative to acidic oxide may be expected to result in more equi-dimensional, fewer and larger crystals. This letter describes such experiments in the system La_2O_3 -PbO-B₂O₃ and experiments in which the proportion of La_2O_3 was varied.

The chemicals used were: 99.99% pure La_2O_3 from Rare Earth Products, BDH "Analar" grade PbO and BDH Laboratory Reagent grade B_2O_3 . The La_2O_3 and B_2O_3 were calcined at 1000 and 700° C, respectively, prior to use.

Many experiments were performed, and Table I reports representative starting compositions, furnace programmes and results. In all cases, the furnace gradient was such that the crucible base was a few degrees cooler than its upper parts.

As the proportion of PbO was increased from 56% to over 80%, the number of crystals decreased from a large number to a few only. With 56% PbO, there were over 50 small platelets about 0.1 mm thick at the melt surface. With over 80% PbO, however, only a few larger plates, up to

2 mm thick, were obtained from a 50 cm^3 crucible. These results are in accord with previous studies in such ternary systems [5, 6].

In other experiments, the concentration of PbO was kept at 80 to 85% while the concentration of La_2O_3 was increased from 3.7 to 5%. Complete solution was obtained at 1250°C for a concentration of 4.6%, and few crystals grew, but at 5% it was evident that solution was incomplete. With 3.7% La2O3, only plates of LaBO3 at the surface were obtained, but with 4.3 and 4.6% La₂O₃, crystals of very different habit grew at the base or lower walls of the crucible. The latter crystals were in the form of (1) tabular prisms (2) rods and (3) approximately equidimensional crystals, as illustrated in Fig. 1. The major faces developed in each type were studied by optical and X-ray techniques. The study showed that the major faces of the platelets are (100), the long directions being the *c*-axis. The tabular prisms have (100) and (110) faces, with the long direction again the c-axis. The rod-shaped crystals have larger (120) faces, and are also elongated in the *c*-direction, with four relatively small facets at each end. The equi-



Figure 2 (a) Platelet. The major faces are (100), the long direction is the *c*-axis. (b) Tabular prism in cross-section, seen along the *c*-axis. (c) Cross-section, viewed along *c*-axis, of a rod. (d) Approximately equi-dimensional crystals (1) viewed along the *c*-axis (a > c). (2) viewed along the *b*-axis, indicating complex morphology.

dimensional crystals show a complex morphology having as many as twenty facets. Fig. 2 shows schematic diagrams in which some of the facets are identified.

Flux growth with relatively high concentrations of refractory oxide and, therefore, at relatively high temperatures, usually favours more equidimensional crystals, as has been noted with Cr_2O_3 , Al_2O_3 , Fe_2O_3 , NiTiO₃, and garnets [7]. Changes in habit similar to those reported here, platy at the melt surface and more equi-dimensional within the melt, have been reported for flux grown MgO, Al_2O_3 , FeVO₄, ZnO, CeO₂ and TiO₂ [7].

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*Electrical conduction in 'pure' and copperdoped KHSO*₄

Electrical conduction in hydrogen-bonded materials such as KH_2PO_4 , $NH_4H_2PO_4$ (both undoped and doped with Ba^{2+} , HSO_4^- and SO_4^{2-} ions), and $(NH_4)_2SO_4:HSO_4^-$ has been demonstrated to be protonic from coulometric determination of transport numbers, electrical conductivity measurements and NMR of deuterated samples [1–5]. Quite recently, KHSO₄ was shown to be a protonic conductor from d.c. electrolysis experiments on Co-doped KHSO₄ crystals and the transport number of protons was found to be nearly unity [6]. We report in this letter, d.c. electrical conductivity results on 'pure' (undoped) and copperdoped KHSO₄ single crystals, in the range 50 to 180° C, the upper temperature limit being dictated by the rather low decomposition temperature (~210° C) of this hydrogen-bonded system. This study was motivated by our earlier work on the ESR of Cu²⁺ impurity in KHSO₄ crystals [7], where we had postulated that both K⁺ and proton vacancies exist in KHSO₄ and that both types